Biosynthesis, Extraction and Characterization of Extracellular Polymeric Substances (EPSs) from *Aspergillus clavatus*

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**Abstract**

Six fungal strains were isolated from El-Sella ore material (Fusarium solani, Penicillium purpurogenum, Aspergillus niger, Aspergillus clavatus, Cladosporium herbarum and Epicoccum nigrum), and screened for exopolysaccharides (EPSs) production. In this study, the most promising fungus producing exopolysaccharides (EPSs) was *A. clavatus* which was identified according to microscopic morphological feature. The effects of environmental parameters on exopolysaccharides production by *A. clavatus* were investigated. The optimum medium composition was found to be at pH 5, glucose and peptone were the best nitrogen and carbon sources. Also, 1 cm disc of inoculum size at 30 °C was incubated for 9 days using shaking condition (150 rpm) were the other optimal conditions. Under these optimum conditions, the predicted yield of EPSs was up to 885.3 mg/L. Further, the structure of exopolysaccharides was confirmed by FT-IR, 1H, 13C NMR, HPLC and mass spectroscopy were employed to characterize the EPSs extracted from *A. clavatus*. Based on this spectroscopy results, the exopolysaccharides produced by *A. clavatus* was α-D-glucopyranosyl units. The HPLC chromatography showed that the EPSs consist of one peak; glucose.

**Keywords**

*Aspergillus clavatus*, EPSs production, Physiological conditions, Characterization, α-D-glucopyranosyl unit

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**Introduction**

Microorganisms can synthesis different type of polymeric substances (PSs) according to their location in the cell, structure, sugar composition, type of bonds between monomers, and systematic affinity. The extracellular polymeric substances (EPSs) are a complex mixture of macromolecular polyelectrolytes including polysaccharides, proteins, nucleic acids [1], amino acids, lipids or humic substances [2]. Fungal PSs are classified into different groups according to systematic affinity, structure (linear and branched), sugar composition (homo- and heteropolysaccharides), type of bonds between the monomers (b-(1-3),b-(1-6), and α-(1-3)) and their location in the cell (cell wall PSs, exo PSs, and endo PSs). Ascomycota and Basidiomycota EPSs are mainly heteropolysaccharides but in the case of homopolysaccharides glucose is their only monomer. Exopolysaccharides are most frequently studied fungal PSs but their definition, classification, and origin are still not clear and should be explained. EPSs are rather defined by the separation/extraction method used than by theoretical consideration of the composition of the cell wall and macromolecules outside the cell wall. Bound or soluble EPSs contain not only high-molecular-weight polymeric compound exudates from microorganisms, but also products of cellular lysis and hydrolysis of macromolecules [3]. Many bacteria and fungi often secrete polysaccharides as an evolutionary adaptation to help them adhere to surfaces and to prevent them from drying out [4]. Ascomycota and Basidiomycota fungal EPSs play different biological functions, for example in the protection against environmental stress factors and in interactions with other organisms. Ascomycota and Basidiomycota fungi are used for biotechnological derivation of EPSs in laboratory conditions using similar techniques, culture conditions, sources of C, N and other elements, culture periods, and mostly acidic pH of the medium [5]. Fungal strains belonging to both these groups are an important component, characterized by the highest biomass, of the microbiome of various environments [6]. Ascomycota and Basidiomycota fungi exhibit a very high variation of the weight of synthesized EPSs and their biochemical and biological properties were varied [7,8]. Many bioactive properties of Ascomycota and Basidiomycota EPSs such as antioxidative, antimicrobial, immunomodulatory, antitumor, hypolipidemic, hypoglycemic, and hepatoprotective activity can find medical applications. The activities of fungal EPSs in such processes as mineral sol-
ubilization, heavy metal sorption and hydrocarbon removal as well as eliciting plant resistance created a possibility of potential environmental and agricultural applications of these EPSs in biofertilization, soil/water bioremediation, and plant bioprotection. The bioactive properties of EPSs are known to depend on many factors, including the structure, monosaccharide components, molecular mass, conformation, configuration of glycosidic bonds, and extraction and isolation methods [9]. Fungal EPSs are derived through ethanol precipitation with different proportions of the culture/water suspension and alcohol [10]. The directions of research on EPSs are mainly focused on identification of factors responsible for their synthesis/release, optimization of production (connected with the cost and productivity).

Materials and Methods

Isolation of fungal strains
Six isolates of fungi namely Penicillium purpurogenum, Fusarium solani, Aspergillus niger, Aspergillus clavatus, Cladosporium herbarum and Epicoccum nigrum were isolated from El-Sella ore material by serial dilution method. Cultures were maintained on Czapek’s agar slants. Slants were inoculated by sporium herbarum sarmiso solani, Aspergillus niger, Aspergillus clavatus, Clado-

Selection of most EPSs producer fungal strain by shake flask culture
Shake flask cultures were carried out in 250 mL Erlenmeyer flasks containing 50 mL of modified Czapek’s broth medium. The media consisting of the following components (in g/L): Sucrose, 30; NaN03, 3.0; KCl 0.5; KH2PO4 0.5; MgSO4•7 H2O 0.5. and Yeast extract, 10. Media were sterilized at 121 °C for 20 min. The pH was adjusted to 6.5. The flasks were incu-
bated on a rotary shaker at 28 °C and 150 rpm for 2 weeks. Then, flasks were estimated for EPSs and biomass mycelial dry weights.

Estimation of mycelia growth and EPSs
Mycelial growth of A. clavatus was expressed as dry cell weight (DCW). The mycelia were harvested at the end of incubation periods and washed with sterile water then filtered through pre-weighed Whatman No.1 filter paper followed by drying at 60 °C for approximately 2 days in an oven to a constant weight, after which DCW were measured.

For total EPSs preparation, the culture broth without mycelia was added with 95% ethanol by two volume times to precipitate total EPSs at 4 °C overnight. After isolation of total EPSs by centrifugation at 7,200 g or 15 minutes, EPSs was resuspended with 95% ethanol and centrifuged again. The total EPSs was dried at 60 °C and the total sugars were deter-

Characterization of extracted exopolysaccharides (EPSs)

Fourier-transform-infrared spectroscopy (FT-IR): FT-IR was used to investigate the vibrations of molecules and polar bonds between the different atoms. Structures of polysaccharides, such as monosaccharides types, glucosidic bonds and functional groups can be analyzed using FT-IR spectroscopy. The sample was pressed into KBr discs. The apparatus used (Model JASCO 4100 FTIR spectrophotometer, Japan) which presented in the Micro Analytical Center, Cairo University, Egypt.

High-performance liquid chromatography (HPLC): HPLC analysis was performed by the HPLC system (Perkin-Elmer, Waltham, MA, USA) with a Perkin-Elmer Brownlee Choice ODS column (150 mm × 4.6 mm i.d.) and UV spectrophotometer detector LC 290. This apparatus was presented in the Mi-cro Analytical Center, Cairo University, Egypt.

Nuclear magnetic resonance (NMR): Proton magnetic resonance (1H-NMR) and proton decoupled 13C nuclear magnetic resonance (13C-NMR) were taken using dimethyl sulfoxide (DMSO) solvent. The exopolysaccaride sample
was dissolved in 0.75 mL DMSO and injected through a 5-mm OD NMR tube to record the nuclear magnetic resonance (1H) (NMR) and (13C) (NMR) at 70 eV and also at room temperature. The 1H-NMR and 13C-NMR spectra were recorded on a Shimadzu GCMS-Qp-1000EX mass spectrometer 300 MHZ which presented in the Faculty of Science, Cairo University, Egypt.

Mass spectroscopy: Mass spectra were obtained using a Shimadzu Qp2010-plus mass spectrometer equipped with an electrostatic and magnetic analyzer (EB part) using quadrupole ion trap (QIT) geometry and an electrospray ionization (ESI II) ion source (Japan); the spray voltage was 70 e.V and the capillary temperature 220 °C. This apparatus was presented in the Regional Center for Mycology and Biotechnology, Azhar University, Cairo, Egypt.

Thermogravimetric analysis (TGA) of EPSs extracted from A. clavatus: Thermal behavior of natural polymer sample was examined by Thermogravimetric Analysis Q500 (TA Instruments). The test was performed in platinum pan, at a temperature from ambient to 800 °C, at heating rate of 10 °C/min under nitrogen purge flow 50 mL/min. This apparatus was presented in Science and Technology Center of Excellence, Cairo, Egypt.

Results and Discussion

Selection of high production exopolysaccharides (EPSs) fungal strain

The purpose of this experiment was to study the best fungal strain which produce high level of EPSs. Six different fungal stains were isolated from El-Sella ore material which namely accordingly to [14,15], *Penicillium purpureogenum*, *Fusarium solani*, *Aspergillus niger*, *Aspergillus clavatus*, *Cladosporium herbarum* and *Epicoccum nigrum*. They were sub-cultured on potato dextrose medium, and then the isolated colonies were grown on potato dextrose broth at 25 °C and incubated for 8-10 days. The availability and stability for extracellular polysaccharides production were tested. It was found that fungus *A. clavatus* is the most promising one for EPSs production and mycelial growth.

The results in Figure 1 appeared that *A. clavatus* was the best strain for the production of EPSs (60.4 mg/50 mL). Also, Di-Hus, et al. [16] revealed that *Ganoderma formosanum* produce about 830.2 mg/L of EPSs under the optimum conditions for maximum production.

Optimization of the fermentation conditions for maximum production of EPSs by *A. clavatus*

Eight factors (Culture medium type, nitrogen sources, carbon sources, pH values, culture types, incubation periods, temperature and inoculum size) were examined independently on the EPSs production *A. clavatus*.

Effect of different culture medium type on EPSs production by *A. clavatus*:

Potato dextrose, Czapek’s dox, modified Czapek’s dox and malt extract media were used to investigate the EPSs production by *A. clavatus* and initial pH was 6 at constant incubation temperature of 25 °C (Figure 2). The maximum EPSs production was achieved in malt extract medium after incubation of 8-10 days in shaking incubator which produced 1.8 g/100 mL of mycelial dry weight and 75 mg/100 mL EPSs. On the other hand Peiqin, et al., [17], who reported that glucose peptone media with different concentration in increased EPSs production of *Berkleasmium sp.*

![Figure 1: Production of EPSs by different fungal strains.](image-url)
Effect of different nitrogen sources on EPSs production by *A. clavatus*: Different nitrogen sources have performed effect on EPSs and biomass production. As illustrated in Figure 3, it is evident that, EPSs concentration using peptone as nitrogen source was 80.3 mg/100 mL and fungal dry weight was 2.1 g/100 mL. The same conclusions were detected by [18] has studied that, the highest level of EPSs was obtained when peptone was used as a sole nitrogen source by *Penicillium commune* KP942881.1.

Effect of different carbon sources on EPSs production by *A. clavatus*: Carbohydrates are a major component of the cytoskeleton and an important nutritional requirement for the growth and development of higher fungi [19]. The effects of carbon sources on the production of EPSs were studied in the fermentation medium containing various carbon sources. The results in Figure 4 showed that glucose is the best carbon source for ESPs production using *A. clavatus* giving 81 mg/100 mL and fungal dry weight was 2.15 g/100 mL followed by sucrose which gave 65 mg/100 mL. It was reported that glucose was the most suitable carbon source for EPSs and mycelium growth production in *Alternaria alternata* [20].

Effect of different pH value on EPSs production by *A. clavatus*: The pH is an important parameter for the efficient production of EPSs. In fact, the pH in fermentation culture control and monitoring can extensively influence yield, cell morphology, chemical composition, and molecular weight of EPSs produced [21]. The study done on the submerged fermentation of *A. clavatus* has shown that the pH for the EPSs synthesis is lower than for the cell growth. The results in Figure 5 appeared that pH 5 is the best one for ESPs production which giving 83.5 mg/100 mL and Mycelial dry weight was 1.9 g/100 mL at pH 7. In contrast to that the curdlan as EPSs...
Figure 4: Effect of different carbon source on the production of EPSs by *A. clavatus*.

Figure 5: Effect of different pH values on the production of EPSs by *A. clavatus*.

Figure 6: Effect of different culture type on the production of EPSs by *A. clavatus*.
of *Alcaligenes faecalis var myxogenes*, the cell growth rate is optimal at pH 7.0, whereas the EPSs production is optimal at pH 5.5 [22].

**Effect of different culture type on EPSs production by *A. clavatus***: Generally, the aeration and agitation are beneficial to the growth and performance (polysaccharide productivity) of the fungal cells by improving the mixing and the heat and mass transfers [23]. From the data presented in Figure 6, shaking condition (150 rpm) was the suitable condition for high mycelial growth production and EPSs of *A. clavatus*. The EPSs concentration in shaking condition was 85 mg/100 mL and mycelial dry weight was 1.9 g/100 mL. So, shaking (150 rpm) condition was selected as the most suitable condition for EPSs production in the subsequent experiments. These results agreed with Kim, et al. [24] have investigated the influence of the aeration rate and agitation intensity on the production of mycelial biomass and EPSs from the entomopathogenic fungus *Paecilomyces sinclairii*.

**Effect of different incubation period on EPSs production by *A. clavatus***: The incubation for 9 days was the most promising for EPSs production and mycelial growth of *A. clavatus*. It was clear that there was a gradual increase in the fungal growth from 5 to 9 days (Figure 7). So, incubation for 9 days was applied for the other experiments. The EPSs concentration was 87 mg/100 mL under shaking condition at 30 °C. Mycelial dry weight was 1.9 g/100 mL. Chen, et al. [25] reported that 10 days was suitable for EPSs production by *Penicillium commune*.

**Effect of different temperature (°C) on EPSs production by *A. clavatus***: The data in Figure 8, it is evident that temperature 30 °C was the optimum for EPSs production and mycelial growth of *A. clavatus*. The EPSs concentration was 87.8 mg/100 mL under shaking condition. Mycelial dry weight was 2.2 g/100 mL. Nehad and Elshamy [20] proved that 30 °C was suitable for EPSs production by *Alternaria alternate*.

**Effect of different inoculum size (disc number) on the production of EPSs by *A. clavatus***: It was observed that three disc sized 1 cm in diameter from *P. commune* gave high EPS dry weight and high mycelial dry weight. Cultivation of three fungal discs with diameter 1 cm from tested organism on malt
extract media at shaking (150 rpm) condition for 9 days gave high quantity of EPSs (88.5 mg/100 mL) and dry mycelial (2 g/100 mL) (Figure 9). Yun, et al. [26] found that inoculum size (5 mm) of Cordyceps militaris was the most favorable for exopolysaccharides production. Osman, et al. [27] observed that increasing the inoculum size led to a significant increase in growth and polysaccharides production.

Under these optimum conditions for EPSs production, the biosynthesis and extraction of it from A. clavatus were evaluated in Figure 10.

Estimation of chemical and elemental compositions of EPSs produced by A. clavatus

Total carbohydrates, reducing sugar, total protein and elemental composition of the EPSs produced by A. clavatus were estimated and the results are presented in Table 1. Total carbohydrates (8.23 mg/100 mL), reducing sugar (1.03 mg/100 mL), protein (2.02 U/mL) were recorded. In addition to, the elemental compositions carbon (24.89%), hydrogen (5.26), nitrogen (1.95%) and C/N (21.76 mol/ratio) were evaluated.

There was a clear trend that EPSs have larger carbohydrates contents; in the present study, in the present study, the reducing sugar, protein and nitrogen contents in EPSs was low as compared to total carbohydrate content. Similar results were also observed by Sharmila, et al. [28] where they recorded that, very low contents of in culture filtrate of the reducing sugar (2.82 g/100 mL), protein (2.30 U/mL) and nitrogen (0.38 g/100 mL) compared to the total carbohydrate content (14.43 g/100 mL).

Characterization of extracted EPSs

Fourier-transform-infrared spectroscopy (FT-IR): The spectrum of the purified EPSs produced by A. clavatus was
As determined (Figure 11). The IR-spectra showed absorption bands and peaks characteristic of carbohydrate, carboxylate, and hydroxyl groups as well as urinate, amine and C-O-C ester linkages. The spectrum of extracted EPSs revealed that a dominant absorption which is often attributed to O-H stretching vibration at 3848.26 cm⁻¹ of O-H in carboxylic acid which is accompanied with the bands at 2932.23 cm⁻¹, known to be typical carbohydrates which corresponds to C-H asymmetric stretching vibration in carboxylic group. The band at 2342.34 cm⁻¹ approves the bending vibration of P-H phosphine group. An asymmetrical stretching peak observed at 1641.13 cm⁻¹ is characteristic of C=C medium stretching vibration in conjugated alkene. The band observed at 1418 cm⁻¹, indicated the presence of urinate. The spectrum also displayed a minor peak at 1379.82 cm⁻¹, indicating the presence of carboxylate in the polymer. The peak at 1238.82 cm⁻¹ identifies the medium vibration stretching of amine C-N in amine group. In addition, the absorption band at 1062.59 cm⁻¹ indicated asymmetrical stretching vibration of a C-O-C ester linkage. The small absorption band at 877.45 cm⁻¹ could be associated with glycosidic linkages between the sugar monomers. The absorption peaks around 1000-1100 cm⁻¹ are known to be characteristic for all sugar derivatives.

The IR-spectrum, in our study, is in consistent with the results reported previously [29]. Thus, the EPSs from A. clavatus contains important functional groups such as carboxyl and hydroxyl. The carboxyl groups present on the molecular chain make the chain stretched-out because of electrostatic repulsion, and the stretched molecular chains provide more effective sites for particle attachment [30]. Furthermore, carboxylate groups act as non-specific ion-exchange material which may convey chelating property [29]. In addition, presence of hydroxyl group within the polymer favored the possibility of hydrogen bonding with one or more water molecules [31]. Presence of urinate assists metal uptake and allows for adhesion of microorganisms to surfaces.

High-performance liquid chromatography (HPLC) analysis of hydrolyzed EPSs-derivatized monosaccharide: The EPSs-derivatized monosaccharide was monitored by an UV detector (Perkin-Elmer) at 245 nm with detector (RI) reflected indict and the flow rate was maintained at 2 mL/min. The sample was eluted with a gradient mobile phase consisting of acetonitrile (75%) and dist. H₂O (25%). The monosaccharide composition of EPSs was determined by the method of sugar using a HPLC analysis technique [32].

For the total hydrolysis of EPSs trifluoroacetic acid (TFA) has important advantages over sulfuric acid. The reaction time is short and there is no need for conventional neutralization, as TFA is volatile and can be removed by evaporation [33]. HPLC analysis suggested that EPSs consisted of only glucose as a monosaccharide unit (Figure 12). This obtained result agreement with that obtained by [34].

Nuclear magnetic resonance (NMR)

Proton nuclear magnetic resonance (¹H NMR): The ¹H NMR spectrum of the EPSs produced from A. clavatus was illustrated in Figure 13 which showed signals of hydrogens that corresponded to the glucosyl residue. The carbonyl hydrogens of D-glucopyranose were observed as a triplet (t) in signals around δ 3.32 to 3.50 ppm. The doublet signals from

<table>
<thead>
<tr>
<th>EPSs chemical compositions</th>
<th>EPSs elemental compositions</th>
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<tr>
<td>Weight of polysaccharides (mg/100 mL)</td>
<td>C (%)</td>
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<tr>
<td>Total carbohydrates (mg/100 mL)</td>
<td>H (%)</td>
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<tr>
<td>Reducing sugar (mg/100 mL)</td>
<td>N (%)</td>
</tr>
<tr>
<td>Protein (U/mL)</td>
<td>C/N mol ratio</td>
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Figure 11: FT-IR spectrum of exopolysaccarides from A. clavatus.
Aspergillus clavatus is a highly linear glucans with α(1→6) and (1→4) glycosidic bonds. All data from NMR experiments (1H and 13C) for the EPSs and commercial D-glucose, as well as the published reports used to confirm the structures of the exopolysaccharides, are summarized in Table 2.

Mass spectrometry: The mass spectrum of EPSs extracted from A. clavatus appeared the fragmentation pattern in Figure 14 with a retention time 3.4 min and relative abundance 98%. The presence of 85.16, 103.24, 149.23, 331.32 and 368.55 m/z ions are inductive that mainly the same of fragmentation of the standard α-D-glucopyranosyl unit. The observed ions in the mass spectrum were also compared with the documented standard of trehalose which consists of 89, 119, 103, 179, 341 and 361 ions Jiang and Coli [36]. From the Figure 15, it was found that, the main product in the produced polysaccharides (EPSs) was consists from α-D-glucopyranosyl unit (C_{12}H_{22}O_{11}) with molecular weight 342 g/moL.
Figure 14: $^{13}$C NMR spectrum of EPSs produced by *A. clavatus* (300 MHz, in DMSO).

Table 2: $^1$H, $^{13}$C NMR data for EPSs produced by *A. clavatus* and literature data [17].

<table>
<thead>
<tr>
<th>Position</th>
<th>EPSs (experimental)</th>
<th>Literature [17]</th>
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<tr>
<td></td>
<td>$^1$H δ</td>
<td>$^{13}$C δ</td>
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<tr>
<td>1</td>
<td>4.88 ppm</td>
<td>100.8 (CH)</td>
</tr>
<tr>
<td>2</td>
<td>3.50 ppm</td>
<td>73.01 (CH)</td>
</tr>
<tr>
<td>3</td>
<td>3.77 ppm</td>
<td>74.13 (CH)</td>
</tr>
<tr>
<td>4</td>
<td>3.53 ppm</td>
<td>78.11 (CH)</td>
</tr>
<tr>
<td>5</td>
<td>3.94 ppm</td>
<td>71.30 (CH)</td>
</tr>
<tr>
<td>6</td>
<td>3.98 ppm</td>
<td>62.41 (CH$_2$)</td>
</tr>
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</table>

Figure 15: Mass spectrum of EPSs produced by *A. clavatus*.

The fragmentation pattern of the second tall peak in Figure 16 with a retention time of 3.3 min and relative abundance 95% is shown in Figure 16. This figure gives the average and monoisotopic residue masses for the common amino
from A. clavatus: TGA technique is important to investigate the thermal stability of natural polymer under heat from ambient up to 800 °C. TG & DTG curves appear in Figure 18. It is clear that the decomposition of the polymer has occurred at three stages as listed in Table 3. The first stage drop starts early up to 200 °C, it was connected with boundary water evaporation and the total moisture content was approx. 32 wt%. The second stage start from 205 to 300 °C, it was starting with dissociation of the polymer chain such as carbon dioxide, carbon monoxide. Finally, the third stage start from 300 to 800 °C, it included the dissociation of the polymer chain releasing different gaseous materials such as acetic acid and crystalline water [38]. The gaseous content was approx. 25.2 wt%.

Conclusion

This present study focused on the production, extraction and characterization of EPSs from A. clavatus. Maximum production of the EPSs from A. clavatus was achieved under shaking condition on malt extract medium at 30 °C and initial pH 5 for 9 days where glucose and peptone were the best carbon and nitrogen sources. The EPSs was purified by successive extraction and precipitation by 95% ethanol. FT-IR spectrum analysis of EPSs indicated the presence of C=O, C-O-C, CH₂ and OH groups and also confirmed the presence of glycosidic linkage. HPLC chromatography showed that the EPSs consisted of one peak; glucose. δHNMR analysis of EPSs confirmed presence of the carbonyl hydrogens of D-glucopyranose (H-4) were observed as a triplet (t) in δ 3.50 ppm. In addition to, The 13C NMR spectrum of EPSs may be confirming that the glucans synthesized by A. clavatus is a highly linear glucans with α (1→6) and (1→4) glycosidic bonds. Finally, acids. It was observed that, 57.08, 72.97, 85.17, 97.15 and 103.11 m/z ions were for glycine, alanine, serine, proline and cysteine. These obtained ions in the mass spectrum were also observed for the mass of an amino acid within a peptide [37].

Based on the FT-IR, HPLC, NMR and mass spectroscopy, it revealed that EPSs produced by A. clavatus containing composed of α-D-glucopyranosyl units (Figure 17) and glucose was the predominant monosaccharides in polysaccharide units.

Other previous study revealed that, polysaccharide fraction could be ascertained with the help of IR-spectroscopy. Based on the TLC, FT-IR and NMR spectroscopy revealed that cell wall of Syncephalastrum sp. contain 3 linked β-D-galactopyronosyl units, and galactose as predominant monosaccharides in polysaccharide units [28].

Thermogravimetric analysis (TGA) of EPSs extracted from A. clavatus: TGA technique is important to investigate the thermal stability of natural polymer under heat from ambient up to 800 °C. TG & DTG curves appear in Figure 18. It is clear that the decomposition of the polymer has occurred at three stages as listed in Table 3. The first stage drop starts early up to 200 °C, it was connected with boundary water evaporation and the total moisture content was approx. 32 wt%. The second stage start from 205 to 300 °C, it was starting with dissociation of the polymer chain such as carbon dioxide, carbon monoxide. Finally, the third stage start from 300 to 800 °C, it included the dissociation of the polymer chain releasing different gaseous materials such as acetic acid and crystalline water [38]. The gaseous content was approx. 25.2 wt%.

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mass spectroscopy for EPSs indicated that polysaccharide (EPSs) was consists from α-D-glucopyranosyl unit \( (C_{12}H_{22}O_{11}) \) with molecular weight 342 g/mol.

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